

Example of Warming

Starting Point

In the container normal room air is at a temperature T of 20°C with a relative humidity j of 50%. Effectively 8.65 g/m^3 of water vapor is present, p at a partial pressure of $1,169 \text{ Pa}$.



Warming 10°C.

The temperature is now 30°C , the quantity of water vapor partial pressure and p remain unchanged. Since the air can now absorb max. 30.4 g/m^3 of water vapor, the relative humidity decreases to 27.6% j



Warming 20°C.

Even at 40°C , 8.65 g/m^3 of water vapor is still present. T , the relative humidity is only 15.8%, p 1,169 Pa remains the same.



Warming 40°C.

Even at 60°C , 8.65 g/m^3 of water vapor is still present

$j = 5.9\%$

$P = 1169 \text{ Pa}$



Heating the air only results in a reduction of j relative humidity from 50% down to 6%.